Tandem Development of Aqueous Indium Chemistry and Ring-Closing Metathesis as a General Route to Fused-Ring α -Methylene- γ -butyrolactones

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Abstract: A program directed toward a general synthesis of α -methylenelactones *cis*- or *trans*-fused to larger rings is reported. The protocol originates with two ω -unsaturated aldehydes of the same or different chain length. One of these is initially transformed by way of the Baylis–Hillman reaction into a functionalized allylic bromide. Merger of the two building blocks is subsequently accomplished in aqueous solution with powdered indium metal serving as the initiator. Once the lactone ring is crafted, the

end products are generated by application of ringclosing metathesis. The central issues surrounding this final step are the effects of the stereochemical disposition of the side chains, the consequences of ring strain, and the locus of the double bonds on cyclization efficiency.

Keywords: alkenes; cyclization; indium; lactones; metathesis

Introduction

The field of synthetic organic chemistry undergoes refinement, gains momentum, and matures as a direct consequence of the discovery of new reagents capable of performing specific chemical transformations efficiently. Organoindium chemistry is an area that has elicited considerable interest in the recent past because of the recognition that selected carbon-carbon bond-forming processes can be implemented in high yield with water as the reaction medium.^[1] These discoveries have had a significant impact on the area of green chemistry and more specifically on the quest for alternative synthetic pathways that result in pollution prevention. Similarly, the development of new reagents capable of accomplishing ring-closing metathesis under notably mild conditions has provided the organic chemist with a powerful tool for molecular constructions.[2]

While both methodologies have *individually* been responsible for numerous advances in synthetic chemistry during the past decade, [3,4] there has to our knowledge appeared no report of their use in a tandem manner. In this context, we saw an opportunity to merge these rapidly expanding technologies so as to achieve a direct, highly convergent route to α -methylene- γ -lactones fused to medium and large rings. [5] Compounds of this general class occur widely in nature. [6] The ability of the unsaturated lactone functionality to act as a highly reactive Michael acceptor has brought forth the realization of its crucial role as a physiologically important

building block.^[7] Often this functionality is found fused to six- to fourteen-membered carbocyclic rings. Included among the many complex structures are euparotin (1),^[8] elephantopin (2),^[9] heliangine (3),^[10] and kericembranolide A (4).^[11]

Understandably, many methods have been developed for proper installation of an α -methylene structural unit onto a γ -lactone ring.^[12] Other more direct approaches

$$\begin{array}{c} H \\ \downarrow \\ H \\ \downarrow \\ H \\ \downarrow \\ A \end{array}$$

Scheme 1.

have also been devised. [13] Nevertheless, a simpler tactic would clearly be desirable. The new strategy is founded on the interrelationship of the structural elements resident in **A** to those present in **B** (Scheme 1). For this tactic to be workable, **B** must be capable of efficient two-fold cyclization, and the interconnective carbon-carbon bond between the upper and lower segments of this hydroxy ester conveniently installed. A notable and important aspect of this retrosynthetic plan is its convergency. Thus, our intent was to craft both building blocks from the same aldehyde **C**.

Results and Discussion

The route starts by Baylis–Hillman coupling^[14] of aldehydes **5** to methyl acrylate in the presence of DABCO (Scheme 2) and progresses forward by conversion of **6** to bromide **7**, a transformation that invariably operates with allylic rearrangement^[15] and gives rise only to the thermodynamically favored Z-isomer.^[16] The condensation of **5** with **7**, under conditions where either m = n or $m \neq n$, is most efficiently carried out with powdered indium metal in 50%

compound	т	n	<i>cis/trans</i> ratio	two-step yield [%]
9a	2	2	3:2	74
9b	3	3	3:2	67
9c	2	7	3:2	40
9d	3	7	3:2	72
9e	7	2	3:2	60
9f	7	3	3:2	70
9g	7	4	3:2	69
9h	7	5	3:2	64

Scheme 2.

aqueous tetrahydrofuran. Good to excellent yields were realized when vigorous stirring was maintained at room temperature for one hour. At high (1 M) bromide concentration, the 1,2-addition leading to 8 proved to be totally γ -regioselective, but not diastereoselective. If In fact, the *cis/trans* ratio in all eight examples was invariably 3:2. In order to facilitate matters, the pairs of hydroxy esters 8 were directly subjected to the action of pyridinium *p*-toluenesulfonate in refluxing benzene in order to effect cyclization to the chromatographically separable *cis*- and *trans*-lactones 9. The distinction between the two isomers was accomplished on the basis of NOE studies. In line with expectation, the effect for *cis*-disposed ring protons was significantly larger than that observed for the more distal *trans* arrangement.

The standardized protocol utilized for all ring-closing metatheses involved 6 mM solutions of 9 in CH₂Cl₂. After the introduction of 30 mol percent of ruthenium catalyst 10 (Figure 1) under N₂, each reaction mixture was heated at 50 °C for 24 h. Under these conditions neither isomer of 9b, 9c, or 9e underwent cyclization, whereas cis-9a was transformed into 11 in only 11% yield at 60% conversion. All of the other substrates examined were smoothly and efficiently converted into cyclized products (Table 1). A mixture of E- and Zisomers was generated in all of the successful case studies, except for the cyclooctene examples 11 and 12. In no instance was there any indication of possible involvement of the conjugated double bond in the metathesis process despite its favorable logistical position in several instances.

For the metathesis products 13 - 20, it was conveniently possible to distinguish the *E*-isomers on the basis of the narrow chemical shift difference exhibited by their two olefinic protons, usually a narrow multiplet near $\delta = 5.5$ in CDCl₃. In contrast, the *Z*-isomers are characterized by two more widely spaced multiplets ($\Delta\delta \approx 0.2$ ppm).^[18]

Table 1. Catalytic ring-closing metathesis results.

substrate	product	E/Z ratio	yield [%]
cis-9a	0 0	0:100	11 ^[a]
H O O	H 0 0	0:100	65 ^[b]

Table 1. Continued.

substrate	product	E/Z ratio	yield [%]
cis-9d	0 0	10:1	68
H 0=0	H O O	4:1	72
trans-9d	14	13:1	72
H O O	H 0 0	7:1	67
cis-9g	pr 0	≔ 0 6:1	71
H O O O	H 0	≔ 0 3:1	70
cis-9h	19	>= 0 3:1	72
HOo	H 0	├ O 3:1	71

[[]a] Under the reaction conditions employed, unreacted cis-9a was recovered to the extent of 39%.

^[b] N. Bensel, H. Marschall, P. Weyerstahl, *Chem. Ber.* **1975**, *108*, 2697.

In line with general precedent, [19] the ring-closing metathesis protocol is not well suited to the formation of cis-fused cyclooctenes. Although cis-9a does not react well, its trans-isomer behaves normally and delivers 12 with the usual efficiency. Therefore, the small increase in ring strain associated with the formation of 11 translates into an unfavorable $K_{\rm eq}$ value for the closure of cis-9a. Medium-sized rings are often more demanding both in terms of catalysts employed and the reaction conditions necessary to achieve good yields. The experimental conditions adopted in this study reflect a good compromise position.

Ring strain contributions do have an effect on the ease of ring closure of bifunctional chain molecules and deserves full consideration in the evaluation of the present results. Galli and Mandolini have examined this matter extensively in general terms and have found that the role of ring strain on the ease of ring closure calls for the strain energies of the transition states to be compared to the strain energies of the ring products. [20] The maximum effects are made evident when the ring size is between 8 and 11. [21,22] Thus, the failure of either isomer of **9b**, **9c**, and **9e** to undergo cyclization is not unexpected in light of the added stereochemical demands brought on by the lactone ring fusion and the increased eclipsing interactions operational in the targeted bicyclic molecules.

A notable feature of those cyclizations that provide 14-membered rings is the more elevated E/Z ratio observed when a *cis*-lactone (e.g., **13** and **15**) is involved. The locus of the double bond in the carbocyclic ring also holds importance. This heightened stereoselectivity is construed to be a reflection of the more elevated barrier to ring closure in the *cis* series due to ring strain. In line with these expectations, the more pronounced mediumring strain effect operational as the ring size is reduced precludes metathesis at the 13-membered ring level independent of cis/trans side chain stereochemistry and the relative positioning of the double bonds (as in 9c and **9e**). At the 15-membered ring level, more equitable E/Zdistributions approaching 3:1 are the norm (as in 17 – 20), indicating that the stereochemistry of the lactone precursor is no longer sterically discriminating.

Conclusion

In summary, a general procedure for the synthesis of α -methylene lactones *cis*- or *trans*-fused to larger rings is described and the existence of certain limitations has been made clear. The convenient approach originates with two ω -unsaturated aldehydes of the same or different chain length. These penultimate intermediates are formed by the application of organoindium chemistry in water. Subsequent to the completion of this investigation, water-soluble metathesis catalysts have been developed^[23] and efficient means for implementing

the Baylis–Hillman reaction in aqueous media have been reported.^[24] Consequently, the prospects for carrying out all three steps in an aqueous environment looms on the horizon.

Experimental Section

General Methods

The column chromatographic separations were performed with Woelm silica gel (230 – 400 mesh). Solvents were reagent grade and in most cases dried prior to use. The purity of all compounds was shown to be > 95% by TLC and high field ¹H (300 MHz) and ¹³C NMR (75 MHz). The high-resolution mass spectra were obtained at The Ohio State University Campus Chemical Instrumentation Center. Elemental analyses were performed at Atlantic Microlab, Inc., Norcross, GA.

Prototypical Baylis-Hillman Coupling: 3-Hydroxy-2-methylenehept-6-enoic Acid Methyl Ester (6a)

A mixture of 4-pentenal (1.91 g, 22.7 mmol), methyl acrylate (3.07 mL, 34.1 mmol) and DABCO (637 mg, 5.68 mmol) was vigorously stirred for 21 days at room temperature. After the removal of excess methyl acrylate under reduced pressure, the residue was chromatographed on silica gel (elution with 20% ethyl acetate in petroleum ether) to furnish **6a** as a colorless oil; yield: 2.37 g (61%). IR (neat): v = 3462, 1718, 1640 cm⁻¹; 1 H NMR (300 MHz, CDCl₃): $\delta = 6.22$ (s, 1H), 5.89 – 5.75 (m, 1H), 5.80 (s, 1H), 5.03 (d, J = 15.6 Hz, 1H), 4.96 (d, J = 10.2 Hz, 1H), 4.40 (dd, J = 12.8, 7.0 Hz, 1H), 3.76 (s, 3H), 2.52 (br s, 1H), 2.27 – 2.06 (m, 2H), 1.82 – 1.64 (m, 2H); 13 C NMR (75 MHz, CDCl₃): $\delta = 166.9$, 142.4, 138.0, 125.0, 115.0, 71.0, 51.8, 35.3, 29.9; HRMS: m/z (M⁺ – H) calcd. 169.0865, obsd. 169.0894.

3-Hydroxy-2-methyleneoct-7-enoic Acid Methyl Ester (6b): Silica gel (elution with 10% ether in petroleum ether); yield: 71%; colorless oil; IR (neat): v = 3456, 1715, 1640 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.18$ (d, J = 0.8 Hz, 1 H), 5.82 – 5.67 (m, 1H), 5.77 (d, J = 0.8 Hz, 1H), 4.96 (d, J = 17.3 Hz, 1H), 4.91 (d, J = 10.2 Hz, 1H), 4.36 (dd, J = 7.0, 5.3 Hz, 1H), 3.73 (s, 3H), 2.69 (br s, 1H), 2.04 (dd, J = 14.4, 7.7 Hz, 2H), 1.69 – 1.34 (m, 4H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 166.9$, 142.5, 138.4, 124.8, 114.5, 71.2, 51.7, 35.6, 33.3, 24.9; HRMS: m/z (M⁺ + H) calcd. 185.1178, obsd. 185.1138; Anal. calcd. for $C_{10}H_{16}O_3$: C 65.19, H 8.75; found: C 65.32, H 8.70.

3-Hydroxy-2-methylenedodec-11-enoic Acid Methyl Ester (6c): Silica gel (elution with 10% ether in petroleum ether); yield: 86%; colorless oil; IR (neat): v = 3462, 1720, 1640 cm⁻¹; 1 H NMR (300 MHz, CDCl₃): $\delta = 6.20$ (d, J = 1.0 Hz, 1H), 5.82 -5.67 (m, 1H), 5.78 (t, J = 1.0 Hz, 1H), 5.01 -4.88 (m, 2H), 4.40 -4.35 (m, 1H), 3.76 (s, 3H), 2.57 (br s, 1H), 2.02 (dd, J = 14.1, 6.8 Hz, 2H), 1.69 -1.54 (m, 2H), 1.49 -1.28 (series of m, 10H); 13 C NMR (75 MHz, CDCl₃): $\delta = 167.0$, 142.5, 139.1, 124.8, 114.0, 71.7, 51.8, 36.2, 33.7, 29.3, 29.0, 28.0, 25.8; HRMS: m/z (M⁺) calcd. 240.1725, obsd. 240.1728. Anal. calcd. for $C_{14}H_{24}O_{3}$: C 69.96, H 10.07; found: C 70.15, H 10.22.

General Bromination Protocol: (Z)-2-Bromomethylhepta-2,6-dienoic Acid Methyl Ester (7a)

To a solution of triphenylphosphine (1.57 g, 5.97 mmol) in dry CH_2Cl_2 (20 mL) was added bromine (310 µL, 5.97 mmol) dropwise at $-20\,^{\circ}\text{C}$. After 15 min, a solution of **6a** (1.00 g, 5.43 mmol) in CH_2Cl_2 (5 mL) was introduced dropwise over 5 min. Following the addition, the cold bath was removed and the reaction mixture was quenched with methanol 30 min later. The solvent was evaporated under reduced pressure and the residue was chromatographed on silica gel (elution with 5% ether in petroleum ether) to afford **7a** as a colorless oil; yield: 1.19 g (68%); IR (neat): v = 1719, 1640, 1440 cm⁻¹; ¹H NMR (300 MHz, C_6D_6): $\delta = 6.95$ (t, J = 7.4 Hz, 1H), 5.87 – 5.73 (m, 1H), 5.10 – 5.00 (m, 2H), 4.21 (s, 2H), 3.78 (s, 3H), 2.39 (dd, J = 14.4, 7.0 Hz, 2H), 2.25 (dd, J = 13.9, 7.0 Hz, 2H); ¹³C NMR (75 MHz, C_6D_6): $\delta = 165.9$, 147.3, 136.7, 129.6, 115.9, 52.1, 31.9, 28.1, 24.1; HRMS: m/z (M⁺) calcd. 232.0100, obsd. 232.0091.

(*Z*)-2-Bromomethylocta-2,7-dienoic Acid Methyl Ester (7b): Silica gel (elution with 5% ether in petroleum ether); yield: 70%; colorless oil; IR (neat): v = 1722, 1641, 1438 cm⁻¹; ¹H NMR (300 MHz, C_6D_6): $\delta = 6.91$ (t, J = 7.7 Hz, 1H), 5.77 – 5.64 (m, 1H), 5.05 – 4.99 (m, 2H), 4.11 (s, 2H), 3.51 (s, 3H), 1.99 (q, J = 7.5 Hz, 2H), 1.91 (q, J = 7.5 Hz, 2H), 1.33 (quintet, J = 7.5 Hz, 2H); ¹³C NMR (75 MHz, C_6D_6): $\delta = 165.6$, 147.6, 138.0, 130.0, 115.3, 51.7, 33.4, 28.2, 27.4, 24.5; HRMS: m/z (M⁺) calcd. 246.0255, obsd. 246.0251.

(*Z*)-2-Bromomethyldodeca-2,11-dienoic Acid Methyl Ester (7c): Silica gel (elution with 5% ether in petroleum ether); yield: 92%; colorless oil; IR (neat): v = 1722, 1641, 1435 cm⁻¹; 1 H NMR (300 MHz, C_6D_6): $\delta = 6.93$ (t, J = 7.6 Hz, 1H), 5.90 – 5.76 (m, 1H), 5.12 – 5.00 (m, 2H), 4.07 (s, 2H), 3.43 (s, 3H), 2.05 – 1.99 (ddd, J = 7.6, 6.8, 6.8 Hz, 2H), 1.93 (q, J = 7.3 Hz, 2H), 1.38 – 1.28 (m, 2H), 1.26 – 1.14 (series of m, 8H); 13 C NMR (75 MHz, C_6D_6): $\delta = 165.7$, 148.0, 139.1, 130.1, 114.6, 51.6, 34.1, 29.5 (2C), 29.2 (2C), 28.8, 28.3, 24.5; HRMS: m/z (M+) calcd. 302.0881, obsd. 302.0903.

NOE difference data for 7c:

$$H_c$$
 H_b H_b H_c H_b H_b

irradiated	δ, ppm	observed	NOE, %
Ha	6.93	H _b	0
H_b	4.07	H _c , H _a	6.4, 0
Me	3.43	H_a,H_c	1.0, 0

Indium-Promoted Coupling of 5 with 7; Direct Conversion to α-Methylene-γ-lactones 9

Bromide **7c** (485 mg, 2.02 mmol) and 5-hexenal (191 mg, 2.22 mmol) were stirred in 2.0 mL of 50% THF/ H_2O for 5 min. Powdered indium metal (253 mg, 2.22 mmol) was introduced and the mixture was vigorously agitated for 1 h. Subsequently, chloroform (40 mL), water (10 mL) and 10% Na_2SO_4 solution (2 mL) were added and the mixture was

stirred for 15 min. The separated aqueous layer was extracted with chloroform (2 × 25 mL) and the combined organic extracts were dried, filtered, and concentrated under reduced pressure. The resulting oil was dissolved in 25 mL of benzene, treated with pyridinium p-toluenesulfonate (100 mg, 0.404 mmol), refluxed for 30 min, and freed of the solvent under reduced pressure. The residue was chromatographed on silica gel with 15% ether in ligroin as eluent to furnish a 3:2 cis/trans mixture of lactones 9d; yield: 423 mg (72%). The isomeric lactones were easily separated on silica gel upon elution with 0-5% ether in ligroin.

cis-3-Methylene-5-(non-8-enyl)-4-(pent-4-enyl)-dihydrofuran-2-one (*cis*-9d): Yield: 43%; colorless oil; IR (neat): ν = 1765, 1640, 1459 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 6.20 (d, J = 2.4 Hz, 1H), 5.87 – 5.72 (m, 2H), 5.50 (d, J = 2.2 Hz, 1H), 5.06 – 4.91 (m, 4H), 4.52 – 4.45 (m, 1H), 2.98 – 2.92 (m, 1H), 2.13 – 2.00 (m, 4H), 1.63 – 1.30 (series of m, 16H); ¹³C NMR (75 MHz, CDCl₃): δ = 170.6, 139.6, 139.1, 137.8, 120.9, 115.2, 114.2, 81.1, 43.1, 33.7, 33.5, 30.2, 29.3, 29.2, 28.9, 28.8, 26.7, 25.9, 25.6; HRMS: m/z (M⁺ – H) calcd. 289.2168, obsd. 289.2164. Anal. calcd. for $C_{19}H_{30}O_2$: C 78.57, H 10.41; found: C 78.67, H 10.52.

NOE difference data for cis-9d:

irradiated	δ, ppm ob	served	NOE, %
Ha	4.52-4.45	H _b	10.9
H_b	2.98-2.92	Ha	11.4

trans-3-Methylene-5-(non-8-enyl)-4-(pent-4-enyl)-dihydro-furan-2-one (*trans*-9d): Yield: 29%; colorless oil; IR (neat): ν = 1766, 1640, 1459 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 6.25 (d, J = 2.6 Hz, 1H), 5.87 – 5.71 (m, 2H), 5.57 (d, J = 2.2 Hz, 1H), 5.05 – 4.91 (m, 4H), 4.16 (dd, J = 10.5, 6.1 Hz, 1H), 2.66 – 2.60 (m, 1H), 2.12 – 2.00 (m, 4H), 1.64 – 1.30 (series of m, 16H); ¹³C NMR (75 MHz, CDCl₃): δ = 170.3, 139.4, 139.1, 137.8, 122.1, 115.2, 114.2, 83.3, 44.4, 36.2, 33.7, 33.5 (2C), 29.3 (2C), 28.9, 28.8, 25.6, 25.1; HRMS: m/z (M⁺ – H) calcd. 289.2168, obsd. 289.2176. Anal. calcd. for C₁₉H₃₀O₂: C 78.57, H 10.41; found: C 78.75, H 10.48.

NOE difference data for trans-9d:

irradiated	δ, ppm	observed	NOE, %
H _a	4.16	H _b	1.9
H _b	2.66-2.60) H _a	1.7

4,5-Di(butenyl)dihydro-3-methylene-2(3*H***)-furanone** (9a): Silica gel (elution with 5 – 10% ether in petroleum ether);

yield: 74%; colorless oil; ¹H NMR (300 MHz, CDCl₃): δ = 6.25 (d, J = 2.6 Hz, 0.42H), 6.20 (d, J = 2.5 Hz, 0.58H), 5.84 – 5.70 (m, 2H), 5.59 (d, J = 2.6 Hz, 0.42H), 5.52 (d, J = 2.5 Hz, 0.58H), 5.10 – 4.98 (m, 2H), 4.50 (q, J = 6.9 Hz, 0.58H), 4.20 (dt, J = 10.6, 6.4 Hz, 0.42H), 3.03 – 2.95 (m, 0.58H), 2.69 – 2.63 (m, 0.42H), 2.36 – 2.01 (series of m, 4H), 1.78 – 1.54 (series of m, 4H). For cis-9a: ¹³C NMR (75 MHz, CDCl₃): δ = 170.3, 139.2, 137.1, 137.0, 121.1, 115.8, 115.7, 80.0, 42.2, 30.4, 29.6, 29.5, 26.4. For trans-9a: ¹³C NMR (75 MHz, CDCl₃): δ = 170.1, 139.0, 137.0, 136.9, 122.4, 115.7, 115.6, 82.4, 43.7, 35.2, 33.2, 30.3, 29.3.

cis-3-Methylene-4,5-di(pent-4-enyl)-dihydrofuran-2-one (*cis*-9b): Silica gel (elution with 5% ether in petroleum ether); yield: 42%; colorless oil; IR (neat): v = 1765, 1640, 1406 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.18$ (d, J = 2.4 Hz, 1H), 5.83 – 5.70 (m, 2H), 5.49 (d, J = 2.4 Hz, 1H), 5.04 – 4.94 (m, 4H), 4.52 – 4.45 (m, 1H), 2.96 – 2.93 (m, 1H), 2.11 – 2.05 (m, 4H), 1.68 – 1.37 (series of m, 8H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.4$, 139.4, 138.0, 137.8, 120.9, 115.2, 115.1, 80.9, 43.1, 33.5, 33.3, 29.5, 26.7, 25.8, 24.8; HRMS: m/z (M⁺) calcd. 234.1620, obsd. 234.1618. Anal. calcd. for C₁₅H₂₂O₂: C 75.88, H 9.46; found: C 77.07, H, 9.35.

NOE difference data for cis-9b:

irradiated	δ, ppm ol	oserved	NOE, %
Ha	4.52-4.45	H _b	8.8
H_b	2.96-2.93	Ha	9.5

trans-3-Methylene-4,5-di(pent-4-enyl)-dihydrofuran-2-one

(*trans*-9b): Silica gel (elution with 5% ether in petroleum ether); yield: 25%; colorless oil; IR (neat): v = 1766, 1637, 1401 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.20$ (d, J = 2.6 Hz, 1H), 5.80 – 5.66 (m, 2H), 5.54 (d, J = 2.6 Hz, 1H), 5.00 – 4.91 (m, 4H), 4.15 – 4.09 (m, 1H), 2.61 – 2.58 (m, 1H), 2.08 – 2.01 (m, 4H), 1.62 – 1.38 (series of m, 8H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.2$, 139.3, 137.9, 137.8, 122.0, 115.2, 115.1, 83.1, 44.4, 35.4, 33.4 (2C), 33.2, 25.5, 24.3; HRMS: m/z (M⁺) calcd. 234.1620, obsd. 234.1633.

NOE difference data for trans-9b:

irradiated	δ, ppm ob	oserved	NOE, %
Ha	4.15-4.09	H _b	1.6
H _b	2.61-2.58	Ha	1.7

cis-4-(But-3-enyl)-3-methylene-5-(non-8-enyl)-dihydrofuran-2-one (*cis*-9c): MPLC on silica gel (elution with 6% ether in petroleum ether); yield: 24%; colorless oil; IR (neat): v = 1766, 1640, 1455 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.21$ (d, J = 2.4 Hz, 1H), 5.87 - 5.72 (m, 2H), 5.52 (d, J = 2.4 Hz, 1H), 5.09 -

4.90 (m, 4H), 4.52 – 4.45 (m, 1H), 3.03 – 2.94 (m, 1H), 2.21 – 2.00 (m, 4H), 1.69 – 1.30 (series of m, 14H); 13 C NMR (75 MHz, CDCl₃): δ = 170.5, 139.4, 139.1, 137.2, 121.1, 115.8, 114.2, 81.0, 42.4, 33.7, 30.5, 30.3, 29.3 (2C), 28.9, 28.8, 26.5, 25.6; HRMS: m/z (M⁺) calcd. 276.2089, obsd. 276.2063. Anal. calcd. for $C_{18}H_{28}O_{2}$: C 78.21, H 10.21; found: C 78.35, H 10.09.

trans-4-(But-3-enyl)-3-methylene-5-(non-8-enyl)-dihydro-furan-2-one (*trans*-9c): MPLC on silica gel (elution with 6% ether in petroleum ether); yield: 16%; colorless oil; IR (neat): v = 1765, 1640, 1453 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.26$ (d, J = 2.5 Hz, 1H), 5.86 - 5.71 (m, 2H), 5.59 (d, J = 2.5 Hz, 1H), 5.08 - 4.90 (m, 4H), 4.18 (dd, J = 10.3, 6.1 Hz, 1H), 2.70 - 2.63 (m, 1H), 2.13 (dd, J = 14.5, 7.4 Hz, 2H), 1.71 - 1.57 (series of m, 4H), 1.55 - 1.25 (series of m, 10H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.2$, 139.2, 139.1, 137.1, 122.3, 115.8, 114.2, 83.3, 43.7, 36.1, 33.7, 33.4, 30.3, 29.2 (2C), 28.9, 28.8, 25.04; HRMS: m/z (M⁺) calcd. 276.2089, obsd. 276.2079. Anal. calcd. for $C_{18}H_{28}O_2$: C 78.21, H 10.21; found: C 78.46, H 9.98.

cis-5-(But-3-enyl)-3-methylene-4-(non-8-enyl)-dihydrofuran-2-one (*cis*-9e): Silica gel (elution with 5 – 10% ether in petroleum ether); yield: 34%; colorless oil; IR (neat): v = 1765, 1660, 1640 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.15$ (d, J = 2.4 Hz, 1H), 5.83 – 5.69 (m, 2H), 5.47 (d, J = 2.4 Hz, 1H), 5.05 – 4.90 (m, 4H), 4.52 – 4.44 (m, 1H), 2.96 – 2.89 (m, 1H), 2.34 – 2.21 (m, 1H), 2.19 – 2.06 (m, 1H), 1.66 – 1.29 (series of m, 16H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.3$, 139.3, 138.8, 137.0, 120.7, 115.6, 114.1, 80.1, 43.0, 33.6, 29.6, 29.4, 29.3, 29.1, 28.8, 28.7, 27.2, 26.4.

trans-5-(But-3-enyl)-3-methylene-4-(non-8-enyl)-dihydro-furan-2-one (*trans*-9e): Silica gel (elution with 5 – 10% ether in petroleum ether); yield: 26%; colorless oil; IR (neat): v = 1763, 1665, 1639 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.23$ (d, J = 2.6 Hz, 1H), 5.85 – 5.71 (m, 2H), 5.56 (d, J = 2.6 Hz, 1H), 5.08 – 4.89 (m, 4H), 4.20 – 4.14 (m, 1H), 2.65 – 2.58 (m, 1H), 2.29 – 2.12 (m, 2H), 2.02 (q, J = 6.9 Hz, 2 H), 1.69 (dd, J = 14.0, 6.6 Hz, 2H), 1.58 – 1.52 (m, 2H), 1.47-1.29 (series of m, 10H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.2$, 139.3, 138.9, 137.0, 122.1, 115.6, 114.2, 82.5, 44.4, 35.3, 34.0, 33.6, 29.4, 29.3, 29.2, 28.9, 28.7, 26.3.

cis-3-Methylene-4-(non-8-enyl)-5-(pent-4-enyl)-dihydro-furan-2-one (*cis*-9f): Silica gel (elution with 5 − 10% ether in petroleum ether); yield: 42%; colorless oil; IR (neat): v = 1767, 1664, 1640, 1459 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.16$ (d, J = 2.5 Hz, 1H), 5.82 − 5.71 (m, 2H), 5.47 (d, J = 2.5 Hz, 1H), 5.02 − 4.88 (m, 4H), 4.48 − 4.45 (m, 1H), 2.94 − 2.91 (m, 1H), 2.09 − 1.98 (m, 4H), 1.66 − 1.29 (series of m, 16H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.4$, 139.4, 139.0, 137.9, 120.7, 114.9, 114.1, 83.8, 43.1, 33.6, 33.2, 33.6, 29.4, 29.1, 28.8, 28.7, 27.2, 26.5, 24.7; HRMS: m/z (M⁺) calcd. 290.2246, obsd. 290.2237. Anal. calcd. for C₁₉H₃₀O₂: C 78.57, H 10.41; found: C 78.50, H 10.31.

NOE difference data for cis-9f:

irradiated	δ, ppm ob	served	NOE, %
Ha	4.48-4.45	H _b	10.5
H_b	2.94-2.91	Ha	12.4

trans-3-Methylene-4-(non-8-enyl)-5-(pent-4-enyl)-dihydrofuran-2-one (*trans*-9f): Silica gel (elution with 5 – 10% ether in petroleum ether); yield: 28%; colorless oil; IR (neat): v = 1766, 1661, 1637, 1460 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.23$ (d, J = 2.5 Hz, 1H), 5.86 – 5.70 (m, 2H), 5.56 (d, J = 2.5 Hz, 1H), 5.04 – 4.89 (m, 4H), 4.18 – 4.12 (m, 1H), 2.64 – 2.57 (m, 1H), 2.12 – 1.99 (m, 4H), 1.66 – 1.29 (series of m, 16H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.3$, 139.4, 139.0, 138.0, 121.9, 115.1, 114.2, 83.2, 44.5, 35.5, 34.1, 33.6, 33.2, 29.4, 29.2, 28.9, 28.8, 26.3, 24.3; HRMS: m/z (M⁺) calcd. 290.2246, obsd. 290.2234. Anal. calcd. for C₁₉H₃₀O₂: C 78.57, H 10.41; found: C 78.71, H 10.33. NOE difference data for *trans*-9f:

irradiated	δ, ppm ob	oserved	NOE, %
H _a	4.18-4.12	H_b	2.1
H_b	2.64-2.57	Ha	1.9

cis-5-(Hex-5-enyl)-3-methylene-4-(non-8-enyl)-dihydrofuran-2-one (*cis*-9g): Silica gel (elution with 5% ether in petroleum ether); yield: 41%; colorless oil; IR (neat): $v = 1767, 1640, 1268 \text{ cm}^{-1}; ^1\text{H NMR } (300 \text{ MHz, CDCl}_3): \delta = 6.16 (d,$ *J*= 2.3 Hz, 1H), 5.84 − 5.70 (m, 2H), 5.48 (d,*J* $= 2.3 Hz, 1H), 5.01 − 4.88 (m, 4H), 4.50 − 4.43 (m, 1H), 2.96 − 2.88 (m, 1H), 2.14 − 2.01 (m, 4H), 1.69 − 1.27 (series of m, 18H); <math>^{13}\text{C NMR } (75 \text{ MHz, CDCl}_3): \delta = 170.4, 139.5, 138.9, 138.4, 120.7, 114.5, 114.1, 81.0, 43.1, 33.6, 33.4, 30.0, 29.4, 29.2, 28.9, 28.7, 28.5, 27.2, 26.5, 25.1; HRMS: <math>m/z$ (M⁺) calcd. 304.2402, obsd. 304.2414. Anal. calcd. for $C_{19}H_{30}O_2$ C 78.90, H 10.59; found: C 78.66, H 10.57.

trans-5-(Hex-5-enyl)-3-methylene-4-(non-8-enyl)-dihydrofuran-2-one (*trans*-9g): Silica gel (elution with 5% ether in petroleum ether); yield: 28%; colorless oil; IR (neat): v = 1766, 1640, 1268 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.23$ (d, J = 2.4 Hz, 1H), 5.85 – 5.70 (m, 2H), 5.55 (d, J = 2.4 Hz, 1H), 5.01 – 4.89 (m, 4H), 4.17 – 4.11 (m, 1H), 2.64 – 2.57 (m, 1H), 2.06 – 1.99 (m, 4H), 1.63 – 1.29 (series of m, 18H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.3$, 139.5, 139.0, 138.4, 121.9, 114.6, 114.2, 83.3, 40.5, 36.0, 34.2, 33.7, 33.5, 29.5, 29.2, 28.9, 28.8, 28.5, 26.3, 24.6; HRMS: m/z (M⁺) calcd. 304.2402, obsd. 304.2401. Anal. calcd. for C₁₉H₃₀O₂: C 78.90, H 10.59; found: C 78.67, H 10.70.

cis-5-(Hept-6-enyl)-3-methylene-4-(non-8-enyl)-dihydro-furan-2-one (*cis*-9h): Silica gel (elution with 5% ether in petroleum ether); yield: 38%; colorless oil; IR (neat): v = 1766, 1640, 1464 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.18$ (d, J = 2.3 Hz, 1H), 5.86 – 5.71 (m, 2H), 5.49 (d, J = 2.3 Hz, 1H), 5.01 – 4.91 (m, 4H), 4.51 – 4.44 (m, 1H), 2.97 – 2.89 (m, 1H), 2.04 – 1.97 (m, 4H), 1.83 – 1.24 (series of m, 20H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.6$, 139.6, 139.0, 138.8, 120.7, 114.3, 114.2, 81.1, 43.2, 33.7, 33.6, 30.2, 29.5, 29.2, 29.0 (2C), 28.8, 28.7, 26.6, 25.5; HRMS: m/z (M⁺) calcd. 318.2559, obsd. 318.2581. Anal. calcd. for C₂₁H₃₄O₂: C 79.19, H 10.76; found: C 79.02, H 10.67.

trans-5-(Hept-6-enyl)-3-methylene-4-(non-8-enyl)-dihydrofuran-2-one (*trans*-9h): Silica gel (elution with 5% ether in petroleum ether); yield: 26%; colorless oil; IR (neat): v = 1766, 1640, 1463 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.24$ (d, J = 2.5 Hz, 1H), 5.87 – 5.56 (m, 2H), 5.02 (d, J = 2.5 Hz, 1H), 4.96 –

4.91 (m, 4H), 4.16 – 4.12 (m, 1H), 2.65 – 2.57 (m, 1H), 2.07 – 2.00 (m, 4H), 1.63 – 0.82 (series of m, 20H); 13 C NMR (75 MHz, CDCl₃): δ = 170.4, 139.5, 139.0, 138.8, 122.0, 114.4, 114.2, 83.4, 44.5, 36.1, 34.2, 33.7, 33.6, 29.5, 29.2, 29.0, 28.8 (2C), 28.7, 26.3, 25.0; HRMS: m/z (M⁺) calcd. 318.2559, obsd. 318.2539. Anal. calcd. for C₂₁H₃₄O₂: C 79.19, H 10.76; found: C 79.04, H 10.68.

General Procedure for Catalytic Ring-Closing Metathesis

A 0.43 mmol sample of the appropriate 9 derivative was placed in a 200-mL round-bottomed flask, sealed with a rubber septum, and flushed with dry N_2 . Dichloromethane (42 mL) and 106 mg (0.129 mmol, 30 mol %) of Grubbs catalyst dissolved in dichloromethane (30 mL) were introduced via cannula under N_2 . After the transfer, the flask was further sealed by placing a second septum (up-side down above the first) and securing it with copper wire. The system was allowed to stir for 24 h at 50 °C, the solvent was evaporated under reduced pressure, and the residue was chromatographed on silica gel using 5% ether/ligroin as the eluent to give the cyclized product. The yields and $\it E/Z$ composition are specified in Table 1.

trans-3-Methylene-3a,4,5,8,9,9a-hexahydro-3*H*-cycloocta[*b*]furan-2-one (12): Colorless oil; ¹H NMR (300 MHz, CDCl₃): $\delta = 6.20$ (d, J = 2.5 Hz, 1H), 5.76 – 5.62 (m, 2H), 5.50 (d, J = 2.5 Hz, 1H); 4.35 – 4.28 (m, 1H), 2.92 – 2.84 (m, 1H), 2.44 – 2.12 (series of m, 4H), 2.10 – 1.94 (m, 2H), 1.56 – 1.43 (m, 2H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 170.0$, 140.5, 129.7, 129.6, 120.7, 83.8, 43.3, 33.2, 30.8, 23.8, 21.5.

cis-3-Methylene-3a,4,5,6,7,8,9,10,13,14,15,15a-dodecahydro-3*H*-1-oxacyclopentacyclotetradecen-2-one (13): Colorless oil; IR (neat): v=1766, 1664, 1443 cm⁻¹; ^{1}H NMR (300 MHz, C_6D_6): $\delta=6.22$ (d, J=2.7 Hz, 0.8H), 6.19 (d, J=2.4 Hz, 0.8H), 5.38-5.28 (m, 0.4H), 5.22-5.14 (m, 0.8H), 5.10-5.01 (m, 0.8H), 5.05 (d, J=2.7 Hz, 0.8H), 5.00 (d, J=2.4 Hz, 0.2H), 3.84-3.79 (m, 0.2H), 3.73-3.68 (m, 0.8H), 2.24-2.18 (m, 1H), 2.06-1.66 (series of m, 4H), 1.53-1.46 (m, 1H), 1.41-0.92 (series of m, 15H); 13 C NMR (75 MHz, C_6D_6) (for major isomer): $\delta=169.5$, 140.2, 132.3, 130.7, 120.6, 81.6, 42.9, 33.5, 31.8, 30.8, 30.0, 27.8, 26.6, 25.9, 25.5, 24.7, 22.7; HRMS: m/z (M+) calcd. 262.1933, obsd. 262.1954.

trans-3-Methylene-3a,4,5,6,7,8,9,10,13,14,15,15a-dodecahydro-3*H*-1-oxacyclopentacyclotetradecen-2-one (14): Colorless oil; IR (neat): v = 1766, 1664, 1442 cm⁻¹; ¹H NMR (500 MHz, C_6D_6): $\delta = 6.22$ (d, J = 2.7 Hz, 0.8H), 6.19 (d, J = 2.4 Hz, 0.2H), 5.38 – 5.28 (m, 0.4H), 5.22 – 5.14 (m, 0.8H), 5.10 – 5.01 (m, 0.8H), 5.05 (d, J = 2.7 Hz, 0.8H), 5.00 (d, J = 2.4 Hz, 0.2H), 3.84 – 3.79 (m, 0.2H), 3.73 – 3.68 (m, 0.8H), 2.24 – 2.18 (m, 1H), 2.06 – 1.66 (series of m, 4H), 1.53 – 1.46 (m, 1H), 1.41 – 0.92 (series of m, 15H); ¹³C NMR (75 MHz, C_6D_6) (for major isomer): $\delta = 169.5$, 140.2, 132.3, 130.7, 120.6, 81.6, 42.9, 33.5, 31.8, 30.8, 30.0, 27.8, 26.6, 25.9, 25.5, 24.7, 22.7; HRMS: m/z (M⁺) calcd. 262.1933, obsd. 262.1954. Anal. calcd for $C_{17}H_{26}O_2$: C 77.82, H 9.99; found: C 77.91, H 9.91.

cis-3-Methylene-3a,4,5,6,9,10,11,12,13,14,15,15a-dodecahydro-3*H*-1-oxacyclopentacyclotetradecen-2-one (15): Colorless oil; IR (neat) v = 1759, 1668, 1442 cm⁻¹; ¹H NMR (500 MHz, C_6D_6): $\delta = 6.16$ (d, J = 3.0 Hz, 1H), 5.27 – 5.20 (m, 0.14H), 5.05 – 4.98 (m, 1.86H), 4.94 (d, J = 3.0 Hz, 1H), 3.94 – 3.90 (m, 0.07H), 2.41 – 2.35 (m, 0.93H), 2.29 – 2.27 (m, 0.07H), 2.01 – 1.65 (series of m, 4H), 1.47 – 0.83 (series of m, 16H); ¹³C NMR (75 MHz, C_6D_6) (for major isomer): $\delta = 169.8$, 140.3, 132.3, 131.5, 118.6,

80.1, 43.2, 31.9, 31.7, 28.7, 27.7, 27.5, 25.9, 25.3, 24.7, 24.3, 23.5; HRMS: *m/z* (M⁺) calcd. 262.1933, obsd. 262.1921.

trans-3-Methylene-3a,4,5,6,9,10,11,12,13,14,15,15a-dodecahydro-3*H*-1-oxacyclopentacyclotetradecen-2-one (16): Colorless oil; IR (neat): v=1766, 1661, 1442 cm⁻¹; ¹H NMR (500 MHz, C₆D₆): $\delta=6.03$ (d, J=3.1 Hz, 1H), 5.22 – 5.17 (m, 0.14H), 5.06 – 5.03 (m, 0.14H), 5.01 – 4.95 (m, 0.86H), 4.90 – 4.79 (m, 0.86H), 4.82 (d, J=3.1 Hz, 1H), 3.67 – 3.64 (m, 0.14H), 3.59 – 3.55 (m, 0.86H), 2.01 – 1.50 (series of m, 5H), 1.37 – 1.23 (series of m, 4H), 1.18 – 0.65 (series of m, 14H); ¹³C NMR (75 MHz, C₆D₆) (for major isomer): $\delta=196.5$, 141.0, 132.6, 130.8, 119.8, 81.7, 44.1, 33.9, 32.0, 31.8, 30.6, 27.4, 26.4, 25.9, 24.7, 24.3, 23.9; HRMS: m/z (M⁺) calcd. 262.1933, obsd. 262.1938.

cis-3-Methylene-3a,5,6,7,8,9,10,13,14,15,16,16a-dodecahydro-3*H*,4*H*-1-oxacyclopentacyclopentadecen-2-one (17): Colorless oil; IR (neat): v = 1766, 1665, 1444 cm⁻¹; ¹H NMR (500 MHz, C_6D_6): $\delta = 6.13$ (d, J = 1.8 Hz, 1H), 5.34 - 5.29 (m, 0.17H), 5.23 - 5.11 (m, 1.83H), 4.97 (d, J = 1.8 Hz, 0.83H), 4.94 (d, J = 2.6 Hz, 0.17H), 4.02 - 3.98 (m, 0.17H), 3.94 - 3.89 (m, 0.83H), 2.36 - 2.32 (m, 0.17H), 2.27 - 2.23 (m, 0.83H), 2.01 - 1.91 (m, 2H), 1.79 - 1.73 (m, 1H), 1.45 - 0.82 (series of m, 19H); 1.79 - 1.79 (m, 19.6), (for major isomer): $\delta = 169.6$, 19.6, 19.

trans-3-Methylene-3a,5,6,7,8,9,10,13,14,15,16,16a-dodecahydro-3*H*,4*H*-1-oxacyclopentacyclopentadecen-2-one (18): Colorless oil; IR (neat): v=1766, 1660, 1460 cm⁻¹; ¹H NMR (500 MHz, C₆D₆): $\delta=6.21$ (d, J=2.4 Hz, 1H), 5.38 – 5.31 (m, 0.25H), 5.26 – 5.21 (m, 0.25H), 5.19 – 4.99 (m, 1.50H), 5.07 (d, J=2.4 Hz, 1H), 3.75 – 3.68 (m, 1H), 2.18 – 2.15 (m, 1H), 2.01 – 1.70 (series of m, 4H), 1.56 – 1.46 (m, 1H), 1.40 – 0.81 (series of m, 17H); ¹³C NMR (75 MHz, C₆D₆) (for major isomer): $\delta=169.3$, 140.9, 131.7, 130.5, 120.9, 82.7, 43.9, 36.0, 33.8, 32.7, 32.2, 28.4, 28.3, 27.6 (2C), 26.7, 25.1, 23.9; HRMS: m/z (M⁺) calcd. 276.2089, obsd. 276.2094. Anal. calcd. for C₁₈H₂₈O₂: C 78.21, H 10.21; found: C 78.17, H 10.27.

cis-3-Methylene-3a,4,5,6,7,8,9,10,13,14,15,16,17,17a-tetradecahydro-3*H*-1-oxacyclopentacyclohexadecen-2-one (19): Colorless oil; IR (neat): v=1759, 1668, 1442 cm⁻¹; ¹H NMR (500 MHz, C_6D_6): $\delta=6.14$ (d, J=3.0 Hz, 1H), 5.39 – 5.37 (m, 0.50H), 5.24 – 5.15 (m, 1.50H), 4.93 (d, J=3.0 Hz, 1H), 4.15 – 4.06 (m, 1H), 2.46 – 2.41 (m, 1H), 2.05 – 1.76 (series of m, 4H), 1.40 – 0.83 (series of m, 20H); ¹³C NMR (75 MHz, C_6D_6) (for major isomer): $\delta=170.8$, 139.5, 131.5, 130.8, 119.8, 80.4, 43.0, 32.3, 31.2, 30.2, 29.1, 28.9, 28.1, 27.8, 27.2, 27.1, 27.0, 24.7, 23.5; HRMS: m/z (M⁺) calcd. 290.2249, obsd. 290.2242. Anal. calcd. for $C_{19}H_{30}O_2$: C 78.57, H 10.41; found: C 78.30, H 10.29.

trans-3-Methylene-3a,4,5,6,7,8,9,10,13,14,15,16,17,17a-tetradecahydro-3*H*-1-oxacyclopentacyclohexadecen-2-one (20): Colorless oil; IR (neat): v=1766, 1660, 1461 cm⁻¹; ¹H NMR (500 MHz, C₆D₆): $\delta=6.21$ (d, J=2.6 Hz, 1H), 5.39 – 5.36 (m, 0.50H), 5.27 – 5.18 (m, 1.50H), 5.00 (d, J=2.6 Hz, 1H), 3.89 – 3.86 (m, 0.75H), 3.81 – 3.78 (m, 0.25H), 2.11 – 1.83 (series of m, 4H), 1.50 – 0.83 (series of m, 20H); ¹³C NMR (75 MHz, C₆D₆) (for major isomer): $\delta=169.4$, 140.8, 131.3 (2C), 120.6, 81.2, 44.0, 36.4, 34.1, 32.8, 32.3, 29.2, 28.7, 28.3, 27.8, 27.7, 26.4, 25.1, 24.8; HRMS: m/z (M⁺) calcd. 290.2249, obsd. 290.2230. Anal. calcd. for C₁₉H₃₀O₂: C 78.57, H 10.41; found: C 78.34, H 10.41.

References and Notes

- [1] a) C.-J. Li, Chem. Rev. 1993, 93, 2023; b) C.-J. Li, T. H. Chan, Organic Reactions in Aqueous Media, John Wiley and Sons, Inc.; New York, 1997; c) L. A. Paquette, in Green Chemistry: Frontiers in Benign Chemical Synthesis and Processing, (Eds.: P. Anastas, T. Williamson), Oxford University Press, 1998; d) L. A. Paquette, in Green Chemistry: Recent Advances in Chemical Processing, (Eds.: P. Anastas, L. Bartlett, T. Williamson), ACS Symposium Series, Vol. 767, 2000, pp. 100-112.
- [2] For recent reviews, consult: a) R. H. Grubbs, S. J. Miller, G. C. Fu, Acc. Chem. Res. 1995, 28, 446; b) H.-G. Schmalz, Angew. Chem. Int. Ed. Engl. 1995, 34, 1833; c) A. Fürstner, Topics Catal. 1997, 4, 285; d) M. Schuster, S. Blechert, Angew. Chem. Int. Ed. Engl. 1997, 36, 2037; e) A. Fürstner, K. Langemann, Synthesis 1997, 792; f) S. K. Armstrong, J. Chem. Soc. Perkin Trans. 1 1998, 371; g) C. Pariya, K. N. Jayaprakash, A. Sarkar, Coord. Chem. Rev. 1998, 168, 1; h) M. L. Randall, M. L. Snapper, J. Mol. Catalysis A: Chem. 1998, 133, 29; i) R. H. Grubbs, S. Chang, Tetrahedron 1998, 54, 4413; j) T. M. Trnka, R. H. Grubbs, Acc. Chem. Res. 2001, 34, 18.
- [3] a) T. H. Chan, C.-J. Li, M. C. Lee, Z. Y. Wei, Can. J. Chem. 1994, 72, 1181; b) A. Lubineau, J. Auge, Y. Queneau, Synthesis 1994, 741; c) C.-J. Li, Tetrahedron 1996, 52, 5643; d) L. A. Paquette, T. M. Mitzel, J. Am. Chem. Soc. 1996, 118, 1931; e) L. A. Paquette, T. M. Mitzel, M. B. Isaac, C. F. Crasto, W. W. Schomer, J. Org. Chem. 1997, 62, 4293; f) L. A. Paquette, G. D. Bennett, A. Chhatriwalla, M. B. Isaac, J. Org. Chem. 1997, 62, 3370; g) M. B. Isaac, L. A. Paquette, J. Org. Chem. 1997, 62, 5333.
- [4] a) R. H. Grubbs, S. Chang, Tetrahedron 1998, 54, 4413;
 b) A. Fürstner, Angew. Chem. Int. Ed. Engl. 2000, 39, 3012;
 c) L. Yet, Chem. Rev. 2000, 100, 2963.
- [5] a) Preliminary communication: L. A. Paquette, J. Méndez-Andino, *Tetrahedron Lett.* 1999, 40, 4031; b) for a more recent example of such merging, see: J. Méndez-Andino, L. A. Paquette, *Org. Lett.* 2000, 2, 1263.
- [6] H. Yoshioka, T. J. Mabry, B. N. Timmermann, Sesquiterpene Lactones, University of Tokyo Press, Tokyo, 1973.
- [7] S. M. Kupchan, D. C. Fessler, M. A. Eakin, T. J. Giacobbe, *Science* 1970, 168, 376.
- [8] a) S. M. Kupchan, J. C. Hemingway, J. M. Cassady, J. R. Knox, A. T. McPhail, G. A. Sim, J. Am. Chem. Soc. 1967, 89, 465; b) S. M. Kupchan, J. E. Kelsey, M. Maruyama, J. M. Cassady, Tetrahedron Lett. 1968, 3517; c) S. M. Kupchan, J. E. Kelsey, M. Maruyama, J. M. Cassady, J. C. Hemingway, J. R. Knox, J. Org. Chem. 1969, 34, 3876.
- [9] S. M. Kupchan, Y. Aynehchi, J. M. Cassady, H. K. Schnoes, A. L. Burlingame, J. Org. Chem. 1969, 34, 3867.
- [10] a) H. Morimoto, Y. Sanno, H. Oshio, *Tetrahedron* 1966, 22, 3173; b) M. Nishikawa, K. Kamiya, A. Takabatake, H. Oshio, Y. Tomiie, I. Nitta, *Tetrahedron* 1966, 22, 3601.
- [11] M. Kobayashi, B. W. Son, Y. Kyogoku, I. Kitagawa, *Chem. Pharm. Bull.* **1986**, *34*, 2306.

- [12] a) N. Petragnani, H. M. C. Ferraz, G. V. J. Silva, Synthesis
 1986, 157; b) H. M. R. Hoffmann, J. Rabe, Angew. Chem. Int. Ed. Engl. 1985, 24, 94; c) P. A. Grieco, Synthesis
 1975, 67; d) R. B. Gammill, C. A. Wilson, T. A. Bryson, Synth. Commun. 1975, 5, 245.
- [13] a) E. Ohler, K. Reininger, U. Schmidt, Angew. Chem. Int. Ed. Engl. 1970, 9, 357; b) L. S. Hegedus, S. D. Wagner, E. Waterman, K. Siiralahansen, J. Org. Chem. 1975, 40, 593; c) Y. Okuda, S. Nakatsukasa, K. Oshima, H. Nozaki, Chem. Lett. 1985, 481; d) H. Shibuya, K. Chashi, K. Kawashima, K. Hori, N. Kurakami, I. Kitagawa, Chem. Lett. 1986, 85; e) J. Nokami, T. Tamaoka, H. Ogawa, S. Wakabayashi, Chem. Lett. 1986, 541; f) K. Uneyama, K. Ueda, S. Torii, Chem. Lett. 1986, 1201; g) J. A. Marshall, S. L. Crooks, B. S. DeHoff, J. Org. Chem. 1988, 53, 1616; h) K. Nishitani, M. Isozaki, K. Yamakawa, Chem. Pharm. Bull. 1990, 38, 28; i) C. Kuroda, S. Anzai, Chem. Lett. 1998, 875.
- [14] a) D. Basavaiah, P. Dharma Rao, R. Suguna Hyma, Tetrahedron 1996, 52, 8001; b) E. Ciganek, Org. React. 1997, 51, 201.
- [15] For an alternative method, see Ref. [12b] and L. A. Paquette, G. D. Bennett, M. B. Isaac, A. Chhatriwalla, *J. Org. Chem.* **1998**, *63*, 1836.

- [16] R. Buchholz, H. M. R. Hoffmann, Helv. Chim. Acta 1991, 74, 1213.
- [17] Total adherence to a Felkin–Anh transition state may be thwarted by additional chelation of indium to the carbomethoxy carbonyl oxygen in the allylindium species. For the latter phenomenon, consult: a) L. A. Paquette, M. B. Isaac, *Heterocycles* **1998**, *47*, 107; b) L. A. Paquette, R. R. Rothhaar, *J. Org. Chem.* **1999**, *64*, 217.
- [18] See, for example: N. A. Porter, V. H.-T. Chang, D. R. Magnin, B. T. Wright, J. Am. Chem. Soc. 1988, 110, 3554.
- [19] S. J. Miller, S.-H. Kim, Z.-R. Chen, R. H. Grubbs, J. Am. Chem. Soc. 1995, 117, 2108.
- [20] C. Galli, L. Mandolini, Eur. J. Org. Chem. 2000, 3117.
- [21] a) H. A. Skinner, G. Pilcher, Quart. Rev. 1963, 17, 264; b) J. B. Pedly, R. D. Naylor, S. P. Kirby, Thermochemical Data of Organic Compounds, 2nd edn., Chapman & Hall, London, 1986; c) J. S. Chicos, D. G. Hesse, S. Y. Panshin, D. W. Rogers, M. Saunders, P. M. Uffer, J. F. Liebman, J. Org. Chem. 1992, 57, 1897.
- [22] N. L. Allinger, M. T. Tribble, M. A. Miller, D. H. Wertz, *J. Am . Chem. Soc.* **1971**, *93*, 1637.
- [23] T. A. Kirkland, D. M. Lynn, R. H. Grubbs, J. Org. Chem. 1998, 63, 9904.
- [24] C. Yu, B. Liu, L. Hu, J. Org. Chem. 2001, 66, 5413.